Docket #: S13-238

# Efficient High-temperature Photoelectrochemical Cell

Stanford engineers have developed an efficient photoelectrochemical cell (PEC) that uses a mixed ion electron conductor (MIEC) heterojunction to enable high temperature (hundreds of <sup>o</sup>C) conversion of concentrated sunlight to chemical fuel (such as hydrogen). At the heart of the solid state PEC is a semiconductor light absorber coated with a thin MIEC layer for improved catalytic activity, electrochemical stability and ionic conduction. This provides a facile path for the ionic carriers to reach the solid electrolyte. This integrated photo-thermochemical device captures both thermal and photon energy to recover solar energy that would otherwise be lost. The single-device, isothermal design is potentially more scalable than more complex conventional thermochemical and hybrid photo-thermochemical water-splitting routes. This technology significantly enhances conversion of solar energy into chemical fuels to help overcome the inherently intermittent nature of solar radiation.



Schematic of elevated-temperature photocathode-based, oxygen-ion conducting PEC (a) with energy band diagram (b). (CB=conduction band and VB=valence band)

#### **Stage of Research**

**Simulation** - predicted solar to hydrogen efficiency of 17% and 11% at 723 and 873<sup>o</sup>K respectively (for an oxygen-ion-conducting photocathode in 1-D with a non-degenerate light absorber with 2.0 eV band-gap and uphill band offset of 0.3 eV)

Experimental - experimental demonstration of this new type of PEC is ongoing

### Applications

- Photoelectrochemical cell (PEC):
  - efficient conversion of solar energy to chemical fuel to enhance total solar energy utilization
  - low cost energy storage in solar power plants

#### Advantages

- Extended operating temperature the MIEC layer allows this new class of PEC to operate at elevated temperature with concentrated solar flux
- Increased efficiency:
  - $\circ\,$  simulations show 17% and 11% at 723 and 873  $^{\rm O}{\rm K}$  respectively
  - $\circ\,$  free energy to dissociate water decreases by 16% from 1.23V at room temperature to 1.04V at 600  $^{\rm o}{\rm C}$
  - good utilization of the solar spectrum
  - thermally enhanced carrier transport, electrocatalysis and fast removal of products in the gas phase decreases probability of carrier recombination
  - $\circ\,$  suppresses dark current
- Single device:
  - integrated photo-thermochemical device captures both thermal and photon energy from concentrated sunlight at temperatures between 673 and 973<sup>o</sup>K
  - potentially more scalable than alternative water-splitting technology
- Isothermal
- No additional energy supply energy for heating reactants is supplied by waste heat

### **Publications**

- Issued US Patent 10,036,093
- Ye X, Melas-Kyriazi J, Feng ZA, Melosh NA, Chueh WC, <u>"A semiconductor/mixed</u> ion and electron conductor heterojunction for elevated-temperature water splitting.", *Phys Chem Chem Phys.* 2013 Oct 7;15(37):15459-69. doi: 10.1039/c3cp52536h.

#### Patents

- Published Application: 20150053568
- Issued: <u>10,036,093 (USA)</u>

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